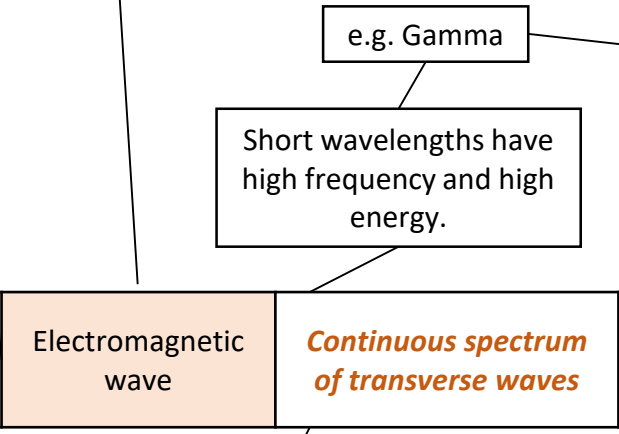


All electromagnetic waves travel at the same velocity in a vacuum: 300 000 000m/s.

Electromagnetic waves are transverse waves that transfer energy from the wave source to an absorber.

	Type	Application	Suitability (HT)
<div>Low frequency low wavelength</div> <div>↑</div> <div>↓</div> <div>High frequency short wavelength</div>	Radio	Television and radio	Travel through atmosphere for long distances
	Microwave	Satellite communications. Cooking food	Travel through atmosphere; agitates water molecules causing them to heat food
	Infrared	Electrical heaters, cooking food, infrared cameras	Heat energy transfer; detection of heat waves
	Visible	Fibre optic communications	Retina can detect light waves; light can travel through optic fibres and carry information
	Ultraviolet	Energy efficient lamps, sun tanning	Some materials can absorb UV and re-emit as visible, energy efficient, skin reacts to UV light causing tanning
	X-rays	Medical imaging and treatment	Pass through soft tissue, penetrate materials to different extents so can produce image
	Gamma rays	Medical imaging and treatment	Kill tissue ; tracers can produce images of internal organs.



Atoms and electromagnetic waves

Input energy could be: light, heat, electricity, X rays etc

Energy out will be a type of electromagnetic radiation i.e. X ray, ultra violet, visible, infra red, microwave or radio waves.

Changes within the nucleus of an atom can result in the emission of gamma waves. This occurs during the radioactive decay of some unstable atoms.

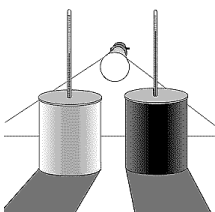
Atoms can receive energy from external sources. This energy can cause electrons to “jump” to a **higher energy level**. When the electron falls back to its original energy level it will release the stored energy in the form of a **photon of electromagnetic radiation**.

P1 Radiation and Waves

What is the Electromagnetic Spectrum?

Black body radiation

Black surfaces	Good emitters, good absorbers
White surfaces	Poor emitters, poor absorbers
Shiny surfaces	Good reflectors

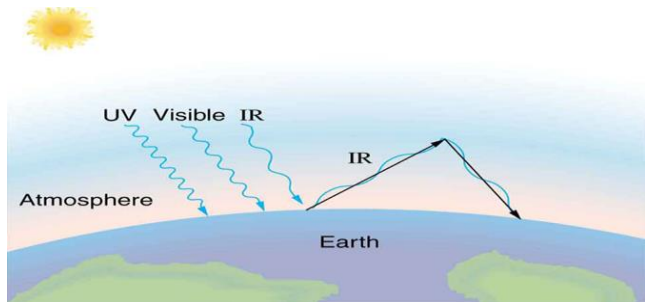


Earth and Global warming

Ultraviolet, visible light, infra-red radiation penetrate atmosphere and heat up Earth's surface.

Longer wavelengths are radiated back, trapped by atmosphere.

Energy lost is not at the same rate as energy being absorbed so Earth heats up.



Black body radiation	All objects absorb or reflect infrared radiation	Hotter objects emit more infrared radiation.
Constant temperature	Rate of absorption = rate of radiation	Intensity and wavelength of energy affects temperature.

EM radiation dangers

Radio wave Dangers	No known dangers
Microwave dangers	Absorbed by water causing it to heat up ☹️ burns under the skin.
Infrared dangers	Surface heating causing burns.
Visible	Damage to eyes
Ionisation	High energy radiation causes ions to form in our cells, damaging DNA and causing cancer.
Ultraviolet dangers	Cell Damage that could lead to skin cancer, snow blindness.
X-ray dangers	Cell Damage that could lead to cancer
Gamma ray dangers	Cell damage that could lead to cancer

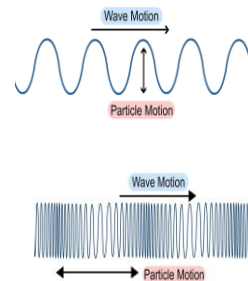
Black surfaces absorb infrared waves better than white or shiny surfaces.



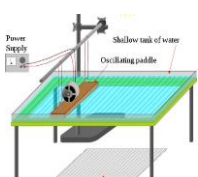
Black surfaces also emit infrared radiation quicker than light coloured surfaces



Wave speed	Wave speed = frequency X wavelength	
Wave period	Wave period = $1 \div \text{frequency}$	$T = 1 \div f$
Speed	Speed = distance ÷ time Calculating wave speed	$v = d \div t$



Transverse wave	Vibration causing the wave is at right angles to the direction of energy transfer	Energy is carried outwards by the wave.	Water and light waves, S waves.
Longitudinal wave	Vibration causing the wave is parallel to the direction of energy transfer	Energy is carried along the wave.	Sound waves, P waves.



In water, use a ripple tank.

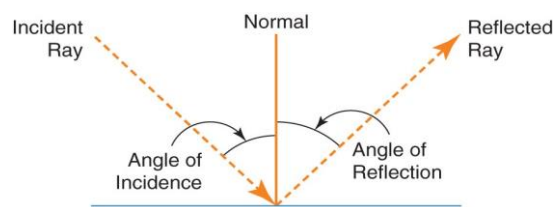
In air, use echoes.

Measuring speed

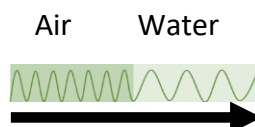
Properties

Waves in air, fluids and solids

Transverse and Longitudinal waves



Angle of incidence = angle of reflection
(i) = (r)



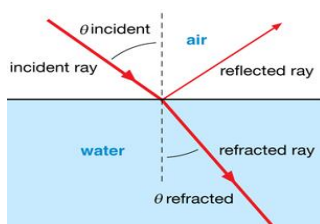
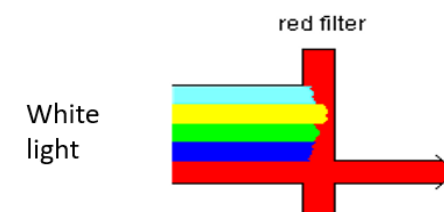
Sound waves travelling through different mediums, the frequency stay constant.

P1 Radiation and Waves How Do Waves Behave?



White light/sunlight is made from all the wavelengths of light in the spectrum. A red object appears red in white light because it only **reflects** the red wavelengths of light, all other colours are absorbed.

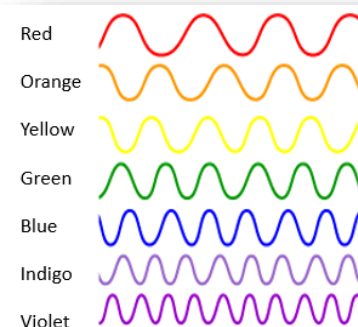
If light **transmits** through a coloured object, the colour passing through is the colour we see. As with reflected light, all other wavelengths of light are absorbed by the transparent or translucent material.



Light refracts as it slows down in a denser substance

Emission	Wave produced by source
Reflection	Wave bounces off the surface.
Refraction	Waves changes direction at boundary.
Transmitted	Passes through the object.
Absorbed	Passes into but not out of, transfers energy and heats up the object.

Waves	Transfer energy without transferring matter.
Oscillate	When particles vibrate backwards and forwards or up and down.
Wavelength	Distance from one point on a wave to the same point of the next wave measured in metres (m)
Amplitude	The maximum disturbance from its rest position
Frequency	Number of waves per second
Period	Time taken to produce 1 complete wave in seconds (s)
Hertz	The unit of frequency. 1 Hz = 1 wave per second.
Velocity	The speed of a wave in m/s.
Medium	The material that waves travel through. Light waves are the only waves that have no medium.



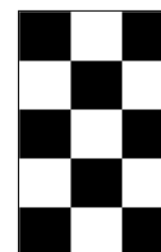
Each colour within the visible spectrum has its own narrow band of **wavelength** and **frequency**.

Colour filters work by **absorbing certain wavelengths** (and colour) and **transmitting other wavelengths** (and colour).

The colour of an opaque object is determined by which wavelengths of light are more **strongly reflected**.

Wavelengths that are **not reflected are absorbed**.

If **all wavelengths are reflected** equally the object will appear **white**. If **all the wavelengths of light are absorbed** equally the object will appear **black**.



Mechanical	<i>Force acts upon an object</i>
Electrical	<i>Electric current flow</i>
Heat	<i>Temperature difference between objects</i>
Radiation	<i>Electromagnetic waves or sound</i>

Energy pathways

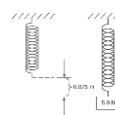
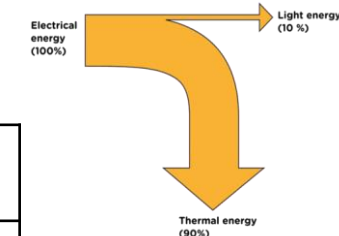
HIGHE: efficiency can be increased using machines.

$$\text{Efficiency} = \frac{\text{Useful power output}}{\text{Total power input}}$$

$$\text{Efficiency} = \frac{\text{Useful output energy transfer}}{\text{Total input energy transfer}}$$

Efficiency *How much energy is usefully transferred*

Useful energy	<i>Energy transferred and used</i>
Wasted energy	<i>Dissipated energy, stored less usefully</i>



Energy stores and changes

P2 Sustainable Energy

Energy Conservation and Dissipation



Dissipate	<i>To scatter in all directions or to use wastefully</i>	When energy is 'wasted', it dissipates into the surroundings as internal (thermal) energy.
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Ways to reduce 'wasted' energy	<i>Energy transferred usefully</i>	Insulation, streamline design, lubrication of moving parts.
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Principle of conservation of energy	<i>The amount of energy always stays the same.</i>	Energy cannot be created or destroyed, only changed from one store to another.
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HIGHER: When an object is moved, energy is transferred by doing work.

$$\text{Work done} = \text{Force} \times \text{distance moved}$$

Frictional forces cause energy to be transferred as thermal energy. This is wasted.

Reducing friction - using wheels, applying lubrication. Reducing air resistance – travelling slowly, streamlining.

Closed system	<i>No change in total energy in system</i>
Open system	<i>Energy can dissipate</i>

Prefix	<i>Multiple</i>	Standard form
Kilo	<i>1000</i>	10^3
Mega	<i>1000 000</i>	10^6
Giga	<i>100 000 000</i>	10^9

Transport	<i>Petrol, diesel, kerosene produced from oil</i>	Used in cars, trains and planes.
Heating	<i>Gas and electricity</i>	Used in buildings.
Electricity	<i>Most generated by fossil fuels</i>	Used to power most devices.

Using renewable energy will need to increase to meet demand.

Renewable energy makes up about 20% of energy consumption.

Fossil fuel reserves are running out.

Energy demand is increasing as population increases.

Non-renewable energy resource	<i>These will run out. It is a finite reserve. It cannot be replenished.</i>	e.g. Fossil fuels (coal, oil and gas) and nuclear fuels.
Renewable energy resource	<i>These will never run out. It is an infinite reserve. It can be replenished.</i>	e.g. Solar, Tides, Waves, Wind, Geothermal, Biomass, Hydroelectric

Using fuels

Energy resources

Generating Electricity

P2 Sustainable Energy

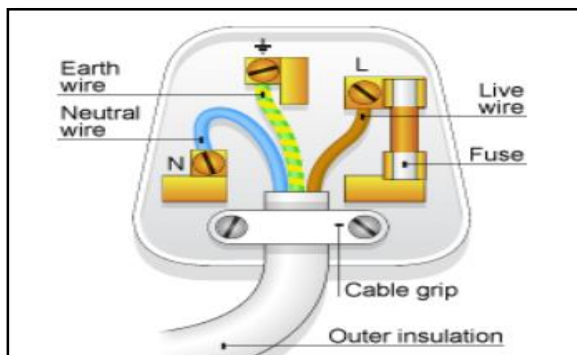
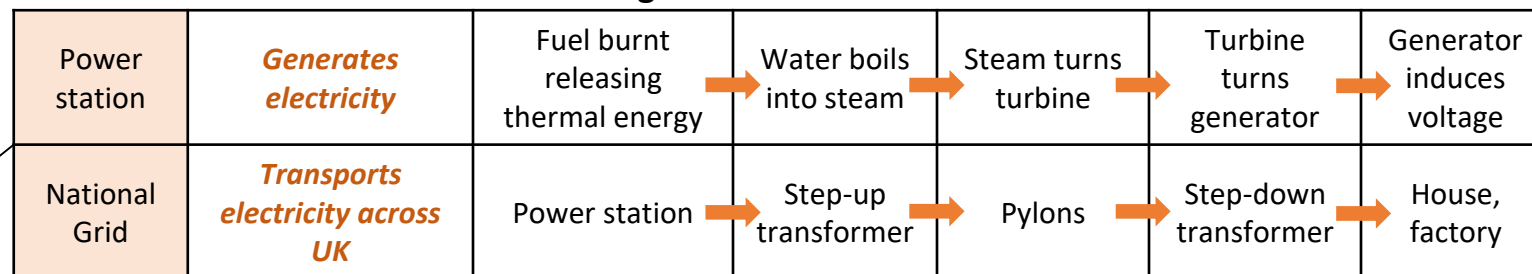
Transformers

Transformers only work with alternating currents ((current flows back and forth)

A lot of energy can be dissipated into the cables as heat which causes a need for transformers

Electricity will travel from live wire to the earth when possible. The Earth is able to absorb electrical charges

Power station – NB: You need to understand the principle behind generating electricity.
An energy resource is burnt to make steam to drive a turbine which drives the generator.



Mains voltage is 230V in the UK and the current is AC at 50Hz

Mains Electricity

P2 Sustainable Energy

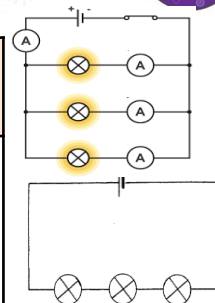
National Grid



Energy resource	<i>How it works</i>	Uses	Positive	Negative
Fossil Fuels (coal, oil and gas)	<i>Burnt to release thermal energy used to turn water into steam to turn turbines</i>	Generating electricity, heating and transport	Provides most of the UK energy. Large reserves. Cheap to extract. Used in transport, heating and making electricity. Easy to transport.	Non-renewable. Burning coal and oil releases sulfur dioxide. When mixed with rain makes acid rain. Acid rain damages building and kills plants. Burning fossil fuels releases carbon dioxide which contributes to global warming. Serious environmental damage if oil spilt.
Nuclear	<i>Nuclear fission process</i>	Generating electricity	No greenhouse gases produced. Lots of energy produced from small amounts of fuel.	Non-renewable. Dangers of radioactive materials being released into air or water. Nuclear sites need high levels of security. Start up costs and decommission costs very expensive. Toxic waste needs careful storing.
Biofuel	<i>Plant matter burnt to release thermal energy</i>	Transport and generating electricity	Renewable. As plants grow, they remove carbon dioxide. They are 'carbon neutral'.	Large areas of land needed to grow fuel crops. Habitats destroyed and food not grown. Emits carbon dioxide when burnt thus adding to greenhouse gases and global warming.
Tides	<i>Every day tides rise and fall, so generation of electricity can be predicted</i>	Generating electricity	Renewable. Predictable due to consistency of tides. No greenhouse gases produced.	Expensive to set up. A dam like structure is built across an estuary, altering habitats and causing problems for ships and boats.
Waves	<i>Up and down motion turns turbines</i>	Generating electricity	Renewable. No waste products.	Can be unreliable depends on wave output as large waves can stop the pistons working.
Hydroelectric	<i>Falling water spins a turbine</i>	Generating electricity	Renewable. No waste products.	Habitats destroyed when dam is built.
Wind	<i>Movement causes turbine to spin which turns a generator</i>	Generating electricity	Renewable. No waste products.	Unreliable – wind varies. Visual and noise pollution. Dangerous to migrating birds.
Solar	<i>Directly heats objects in solar panels or sunlight captured in photovoltaic cells</i>	Generating electricity and some heating	Renewable. No waste products.	Making and installing solar panels expensive. Unreliable due to light intensity.
Geothermal	<i>Hot rocks under the ground heats water to produce steam to turn turbine</i>	Generating electricity and heating	Renewable. Clean. No greenhouse gases produced.	Limited to a small number of countries. Geothermal power stations can cause earthquake tremors.
better hope – brighter future				

Electrons carry current.
Electrons are free to move in metal.

Cell	Battery	Switch	Lamp	Ammeter	Volt meter	Diode	LED	LDR	Fuse	Resistor	Variable resistor	Thermistor
<i>Store of chemical energy</i>	<i>Two or more cells in series</i>	<i>Breaks circuit, turning current off</i>	<i>Lights when current flows</i>	<i>Measures current</i>	<i>Measures potential difference</i>	<i>Current flows one way</i>	<i>Emits light when current flows</i>	<i>Resistance low in bright light</i>	<i>Melts when current is too high</i>	<i>Affects the size of current flowing</i>	<i>Allows current to be varied</i>	<i>Resistance low at high temp</i>



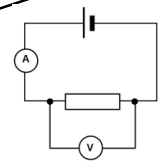
Current	<i>Flow of electrical charge</i>	Ampere (A)
Potential difference (p.d.)	<i>How much electrical work is done by a cell</i>	Volts (V)
Charge	<i>Amount of electricity travelling in a circuit</i>	Coulombs (C)

Charge = Current X time

$Q = I \times t$

Changing current	<i>Change the p.d. of the cells</i>
	<i>Add more components</i>

Controlling current

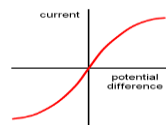
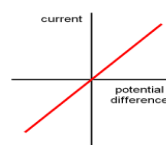


$R = V \div I$

Resistance = Potential difference ÷ Current

Ammeter	<i>Set up in series with components</i>
Voltmeter	<i>Set up parallel to components</i>

Resistance (Ω)	<i>A measurement of how much current flow is reduced</i>
The higher the resistance, the more difficult it is for current to flow.	
Increasing resistance, reduces current.	
Increasing voltage, increases current.	



Ohmic conductor	<i>At a constant temperature, current is directly proportional to the p.d. across the resistor.</i>
Filament lamp	<i>As current increases, the resistance increases. The temperature increases as current flows.</i>
Diode	<i>Current flows when p.d. flows forward. Very high resistance in reverse.</i>

Current – Potential difference graphs

Circuit symbols

Current and Charge

Current, potential difference and resistance

Series and parallel circuits

P3 Electric Circuits

Domestic uses and safety

Series circuit	Current is the same in all components.	Total p.d. from battery is shared between all the components.	Total resistance is the sum of each component's resistance.
Parallel circuit	Total current is the sum of each component's current.	p.d. across all components is the same.	Total resistance is less than the resistance value of the smallest individual resistor.

Series	Parallel
<i>A circuit with one loop</i>	<i>A circuit with two or more loops</i>
Total p.d.	<i>If cells are joined in series, add up individual cell values</i>

Energy transfers

Power (W) = potential difference X current

$R = V \times I$

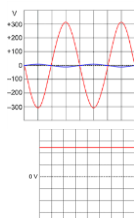
Power = (current)² X resistance

$P = I^2 \times R$

Energy transferred = Power X time

$E = P \times t$

Work is done when charge flowing.



National Grid

Distributes electricity generated in power stations around UK

Step-up transformers	Step-down transformers
<i>Increase voltage, decrease current</i>	<i>Decrease voltage, increase current</i>
Increases efficiency, reduces heat loss.	Makes safer for houses.

Static electricity

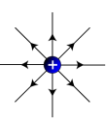
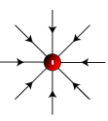
PHYSICS only

Static electricity	<i>Electrical charge is stationary</i>	When two insulating material are rubbed together, electrons move from one material to the other.
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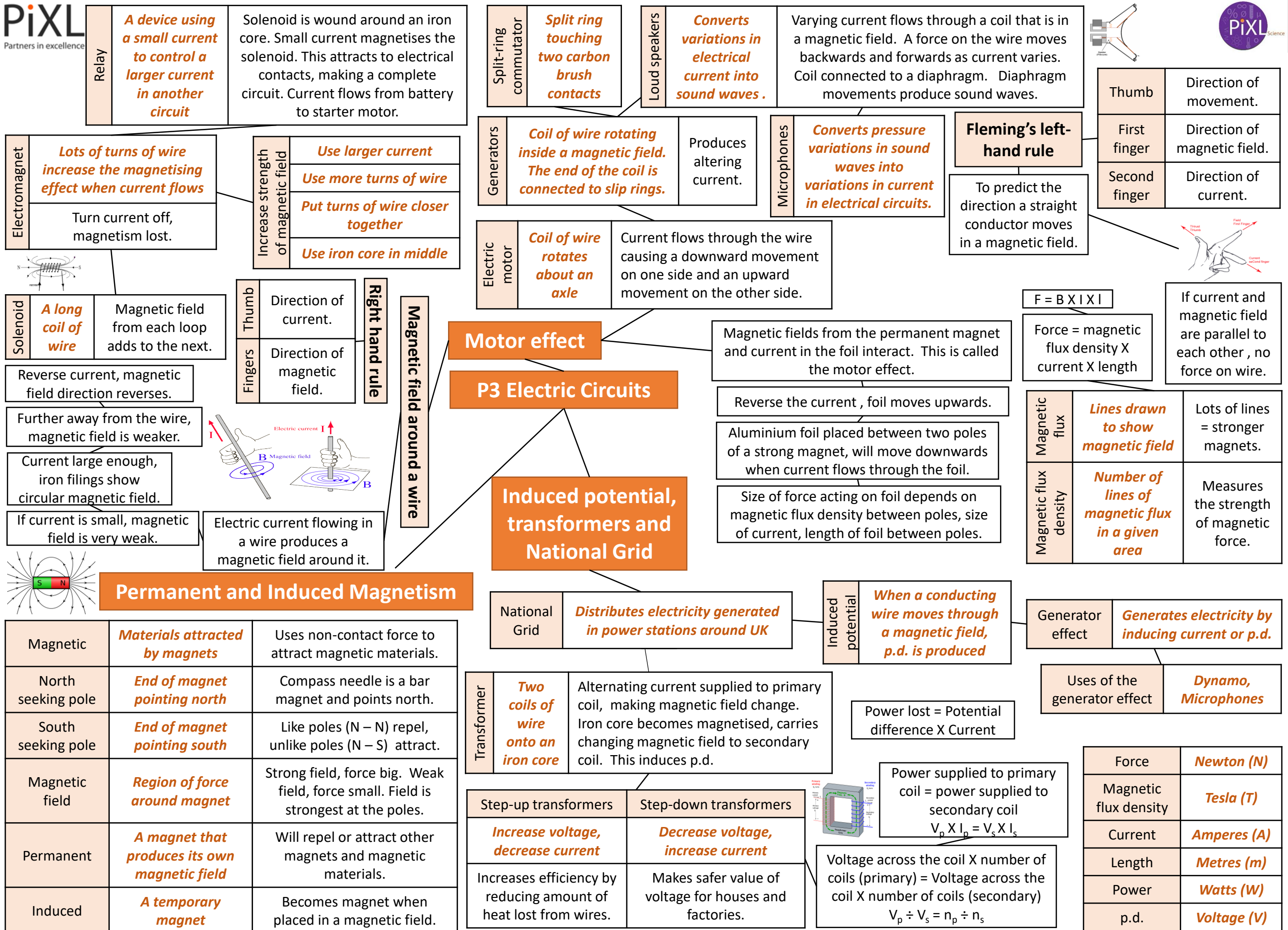
Shocks
Walking on carpet causes friction. Electrons move to the person and charge builds up. When the person touches a metal object, the electrons conduct away, making a spark.

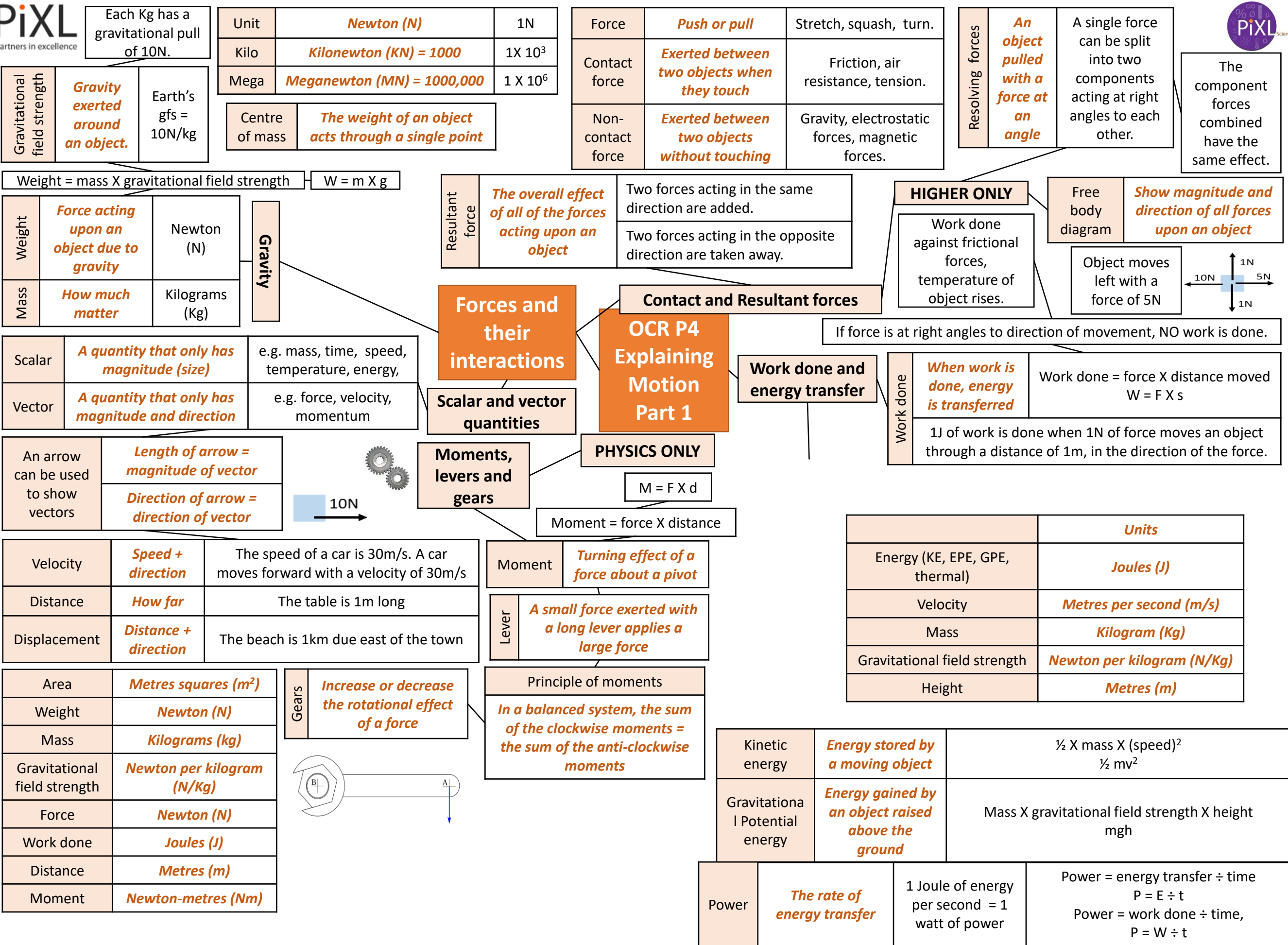
Electric fields
Charged objects create electric fields around them. Strongest closest to the object. The field direction is the direction of force on a positive charge. Add more charge increases field strength.

Like charges	<i>Repel</i>
Unlike charges	<i>Attract</i>



Mains supply
Frequency 50Hz, 230V





Aeroplane banks to change direction	Velocity changes.
Car travelling around a bend	Constant speed, direction changes.
Satellite orbiting the Earth	Constant speed, direction changes.

Distance travelled **Area under the graph shape**

Constant acceleration
 $(\text{final velocity})^2 - (\text{initial velocity})^2 = 2 \times \text{acceleration} \times \text{distance}$
 $V^2 - u^2 = 2 \times a \times s$

Gradient = vertical \div horizontal

HIGHER ONLY

Accelerating objects
It takes time for objects to reach top speed
Draw a tangent to the curve, work out gradient.

Velocity-time graph
Shows speed of an object

Accelerating **Object getting faster**
Decelerating **Object slowing down**

Falling objects

Falling objects accelerate due to gravity.
In no air resistance, objects accelerate at 9.8m/s^2
Air resistance slows falling objects down.

Terminal velocity
Weight of an object is balanced by resistive forces
Object moves at a constant velocity. Resultant force = 0.

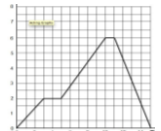
Parachuting
Size of air resistance depends on area of object and speed

Larger the area, the larger the air resistance.
Larger the speed, the larger the air resistance.

Inertia
When objects continue in the same state of motion
Speed or direction only changes if a resultant force acts on the object

HIGHER ONLY

Forces, acceleration and Newton's Laws of motion



Distance-time graph
Shows how far an object moves along a straight line
Speed of object
Use the gradient of graph

Describing motion

Speed is rarely constant.

OCR P4 Explaining Motion Part 2

Observing and recording motion

Force = mass \times acceleration

HIGHER ONLY

$F = m \times a$

Inertial mass
How difficult it is to change the velocity of an object

Inertial mass = force \div acceleration
If the mass is large, to change velocity a big force is needed.

Acceleration is proportional to resultant force.
Acceleration is inversely proportional to mass.

Newton's first Law	Balanced forces	When the resultant force on an still object = 0, the object is stationary.
Newton's second Law	Unbalanced forces	When the resultant force on a moving object = 0, the object is at a constant speed.
Newton's third Law	Equal and opposite forces	When the resultant force is greater than 0, the object accelerates. It could speed up, slow down or change direction.
		When two objects interact the forces exerted are equal and in an opposite direction.

Momentum

HIGHER ONLY

Is a vector

$p = m \times v$

Momentum = mass \times velocity

Conservation of momentum
When two objects collide, the momentum they have before the collision = the momentum they have after the collision
Closed system = no external forces acting on it.

HIGHER ONLY

Changes in momentum
Force is applied to stop momentum
If momentum changes slowly, the force applied is small so less damage.

Crumple zones

Speed / velocity	Metres per second (m/s)
Distance	Metres (m)
Time	Seconds (s)
Acceleration	Metres per second squared (m/s²)
Force	Newton (N)
Mass	Kilogram (Kg)
Momentum	Kilograms metres per second (Kgm/s)

Changing velocity
Objects in a circular motion, change direction but keep a constant speed

Velocity
The speed of an object with direction
Vector

HIGHER ONLY

Speed of sound 330m/s .

HIGHER ONLY

Speed = distance \div time
 $v = s \div t$

Speed	How fast an object moves	Scalar
Displacement	Includes the distance and direction an object moves	vector
Distance	How far an object moves	scalar

Car on motorway	30m/s	Walking	1.5m/s
Train	60m/s	Running	3m/s
Jet plane	200m/s	Cycling	6m/s

Speed affects both thinking and braking distances.

Typical reaction time = 0.7s

Frictional forces decelerate a moving object and bring it to rest.

Forces and braking

Thinking distance	Distance travelled whilst the driver reacts
Braking distance	Distance travelled whilst the car is stopped by the brakes
Stopping distance	Total thinking and braking distances

Factors affecting stopping distances	Drivers reaction times	Drinking alcohol, taking drugs, tired.
	Braking distances	Weather conditions, worn brakes or tyres, road surface, size of braking force.

Braking and kinetic energy	Work done by braking force, reduces kinetic energy	Kinetic energy decreases, temperature of brakes increases due to frictional forces.
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Radius of an atom
 $1 \times 10^{-10}\text{m}$



Electrons gained
Negative ion

Electrons lost
Positive ion

Atom	Same number of protons and electrons
Ion	Unequal number of electrons to protons
Mass number	Number of protons <u>and</u> neutrons
Atomic number	Number of protons

Particle	Charge	Size	Found
Neutron	None	1	In the nucleus
Proton	+	1	
Electron	-	Tiny	Orbits the nucleus

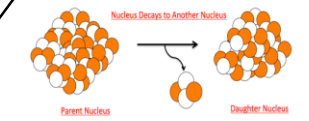
Isotope

${}^6_3\text{Li}$
 ${}^7_3\text{Li}$

Different forms of an element with the same number of protons but different number of neutrons

Discovery of the nucleus

Democritus	Suggested idea of atoms as small spheres that cannot be cut.
J J Thomson (1897)	Discovered electrons– emitted from surface of hot metal. Showed electrons are negatively charged and that they are much less massive than atoms.
Thomson (1904)	Proposed ‘ <i>plum pudding</i> ’ model – atoms are a ball of positive charge with negative electrons embedded in it.
Geiger and Marsden (1909)	Directed beam of alpha particles (He^{2+}) at a thin sheet of gold foil. Found some travelled through, some were deflected, some bounced back.
Rutherford (1911)	Used above evidence to suggest alpha particles deflected due to electrostatic interaction between the very small charged nucleus, nucleus was massive. Proposed mass and positive charge contained in nucleus while electrons found outside the nucleus which cancel the positive charge exactly.
Bohr (1913)	Suggested modern model of atom – electrons in circular orbits around nucleus, electrons can change orbits by emitting or absorbing electromagnetic radiation. His research led to the idea of some particles within the nucleus having positive charge; these were named protons.
Chadwick (1932)	Discovered neutrons in nucleus – enabling other scientists to account for mass of atom.



Radioactive decay	Unstable atoms randomly emit radiation to become stable
Detecting	Use Geiger Muller tube
Unit	Becquerel
Ionisation	All radiation ionises

Atoms and Isotopes

Atoms and Nuclear Radiation

OCR P5

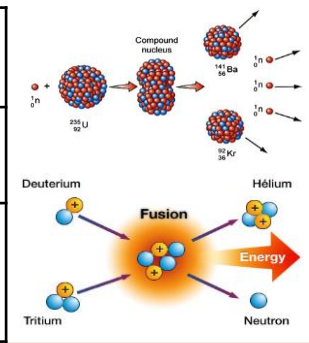
PHYSICS ONLY: Hazards and uses of Radioactive emissions and of background radiation

Nuclear fission and fusion

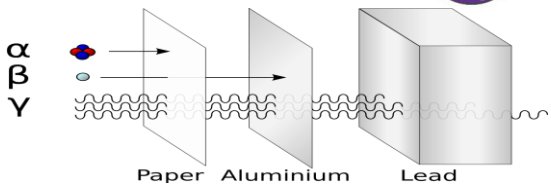
PHYSICS ONLY: Nuclear energy

Fuel rods	Made of U-238, ‘enriched’ with U-235 (3%). Long and thin to allow neutrons to escape, hitting nuclei.
Control rods	Made of Boron. Controls the rate of reaction. Boron absorbs excess neutrons.
Concrete	Neutrons hazardous to humans – thick concrete shield protects workers.

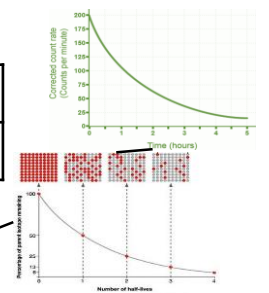
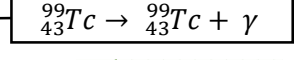
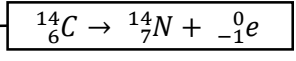
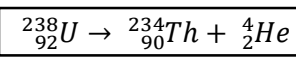
Nuclear fission	One large unstable nucleus splits to make two smaller nuclei	Neutron hits U-235 nucleus, nucleus absorbs neutron, splits emitting two or three neutrons and two smaller nuclei. Process also releases energy.	Process repeats, chain reaction formed
Nuclear fusion	Two small nuclei join to make one larger nucleus	Difficult to do on Earth – huge amounts of pressure and temperature needed.	Occurs in stars



Decay	Range in air	Ionising power	Penetration power
Alpha	Few cm	Very strong	Stopped by paper
Beta	Few m	Medium	Stopped by Aluminium
Gamma	Great distances	Weak	Stopped by thick lead



Decay	Emitted from nucleus	Changes in mass number and atomic number	
Alpha (α)	Helium nuclei (${}^4_2\text{He}$)	-4	-2
Beta (β)	Electron (${}^0_{-1}\text{e}$)	0	+1
Gamma (γ)	Electromagnetic wave	0	0
Neutron	Neutron	-1	0

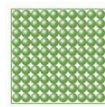
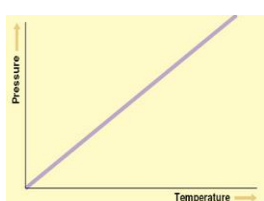


Contamination	Unwanted presence of radioactive atoms
Irradiation	Person is in exposed to radioactive source

Half life	The time taken to lose half of its initial radioactivity
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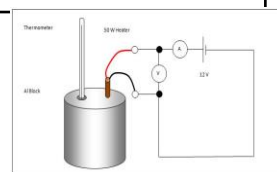
Sievert	Unit measuring dose of radiation
Background	Constant low level environmental radiation, e.g. from nuclear testing, nuclear power, waste

Uses	Different isotopes have different half lives	Short half-lives used in high doses, long half lives used in low doses.
Tracers	Used within body	Isotope with short half life injected, allowed to circulate and collect in damaged areas. PET scanner used to detect emitting radiation. Must be beta or gamma as alpha does not penetrate the body.
Radiation therapy	Used to treat illnesses e.g. cancer	Cancer cells killed by gamma rays. High dose used to kill cells. Damage to healthy cells prevented by focussed gamma ray gun.



State	Particle arrangement	Properties
Solid	Packed in a regular structure. Strong forces hold in place so cannot move.	Difficult to change shape.
Liquid	Close together, forces keep contact but can move about.	Can change shape but difficult to compress.
Gas	Separated by large distances. Weak forces so constantly randomly moving.	Can expand to fill a space, easy to compress.

	Units
Density	Kilograms per metre cubed (kg/m³)
Mass	Kilograms (kg)
Volume	Metres cubed (m³)
Energy needed	Joules (J)
Specific latent heat	Joule per kilogram (J/kg)
Change in thermal energy	Joules (J)
Specific heat capacity	Joule per kilogram degrees Celsius (J/kg°C)
Temperature change	Degrees Celsius (°C)
Pressure	Pascals (Pa)



PHYSICS ONLY: when you do work the temperature increases e.g. pump air quickly into a ball, the air gets hot because as the piston in the pump moves the particles bounce off increasing kinetic energy, which causes a temperature rise.

Reducing the volume of a fixed mass of gas increases the pressure.

Halving the volume doubles the pressure.

$$PV = \text{constant.}$$

$$P_1 V_1 = P_2 V_2$$

Change in thermal energy = mass **X** specific heat capacity **X** temperature change.

$$\Delta E = m \times c \times \Delta \theta$$

Depends on:

- Mass of substance
- What the substance is
- Energy put into the system.

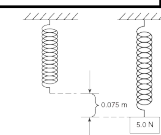
Specific Heat Capacity

Energy needed to raise 1kg of substance by 1°C

OCR P6 Matter – models and explanations

Internal energy and energy transfers

Internal energy	Energy stored inside a system by particles	Internal energy is the total kinetic and potential energy of all the particles (atoms and molecules) in a system.
	Heating changes the energy stored within a system	Heating causes a change in state. As particles separate, potential energy stored increases. Heating increases the temperature of a system. Particles move faster so kinetic energy of particles increases.



Energy needed = mass **X** specific latent heat.

$$\Delta E = m \times L$$

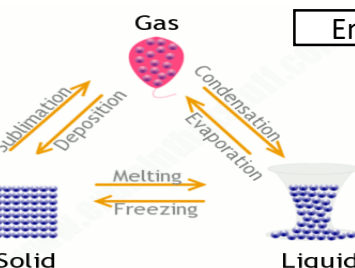
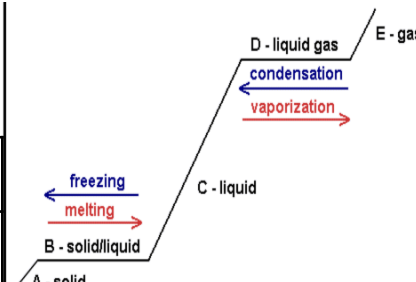
Stretching a spring

Force = spring constant **X** extension, $F = k \times e$

$$EPE = \frac{1}{2} \times \text{spring constant} \times (\text{extension})^2, EPE = \frac{1}{2} k e^2$$

Elastic Potential energy (EPE)

Energy stored in a stretched spring



Kinetic theory of gases

Particle model

Pressure of a fixed volume of gas increases as temperature increases (temperature increases, speed increases, collisions occur more frequently and with more force so pressure increases).

Temperature of gas is linked to the average kinetic energy of the particles.

If kinetic energy increases so does the temperature of gas.

No kinetic energy is lost when gas particles collide with each other or the container.

Gas particles are in a constant state of random motion.

$$P = m \div V$$

Density = mass \div volume.



Density

Mass of a substance in a given volume

Freezing	Liquid turns to a solid. Internal energy decreases.
Melting	Solid turns to a liquid. Internal energy increases.
Boiling / Evaporating	Liquid turns to a gas. Internal energy increases.
Condensation	Gas turns to a liquid. Internal energy decreases.
Sublimation	Solid turns directly into a gas. Internal energy increases.
Conservation of mass	When substances change state, mass is conserved.
Physical change	No new substance is made, process can be reversed.



Milky Way
our galaxy.

Planet	<i>A large body orbiting the Sun</i>
Moon	<i>A natural satellite orbiting a planet</i>
Dwarf planet	<i>A body large enough to have its own gravity which caused a spherical shape</i>
Solar system	<i>Any object orbiting the Sun due to gravity</i>
Galaxy	<i>Collection of billions of stars</i>
Universe	<i>Collection of galaxies</i>



Comets, asteroids, satellites.

Other objects.

Solar system

Effect of gravity.

Gravity causes moons to orbit planets, planets to orbit the Sun, stars to orbit galaxy centres.

Force of gravity changes the moon's direction not its speed.

Gravity pulls objects towards the ground.

Speed of Orbit.

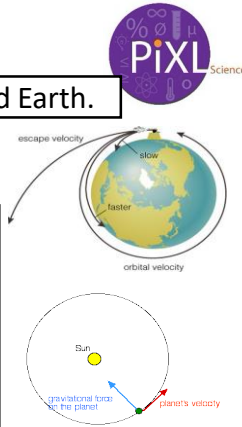
Orbital motions

Too fast = disappears into Space.

Correct speed = steady orbit around Earth.

Too slow = falls to Earth.

To calculate speed of Orbit: distance object moves in 1 orbit, Distance = $2\pi r$, then average speed = distance \div time.



HIGHER: Circular orbits.

Velocity = a vector.

A planet's velocity changes but speed remains constant.

Due to the Sun's gravity, planets accelerate towards the Sun and so changes direction.

When ambulances go past the sound changes from a high pitch to a low pitch.

Planets close to the Sun, gravity pull is strong. Planets move quickly.

Planets further away from the Sun, gravity pull is weaker. So speed of planet is slower.

Frequency of sound wave decreases, wavelength increases.

The life cycle of a star.

Nebula	<i>A cloud of cold hydrogen gas and dust</i>	Cloud collapses due to gravity, particles move very fast colliding with each other, kinetic energy transfers into internal energy and the temperature increases.
Protostar	<i>The large ball of gas contracts to form a star</i>	High temperature causes Hydrogen nuclei to collide and nuclear fusion begins. A star is 'born'.
Main sequence	<i>Stable period of star</i>	Gravity tries to collapse the star but enormous pressure of fusion energy expands and balances the inward force.



Stars the same size as our Sun.

Red giant	<i>A large star that fuses Helium into heavier elements</i>	Hydrogen runs out, star becomes unstable, pressure inside drops causing star to collapse. Atoms now closer together results in atoms fusing and temperature increases. This increase in temperature causes the core to swell.
White dwarf	<i>Star collapses</i>	Nuclear fuel runs out, fusion stops, dense very hot core.
Black dwarf	<i>Cold dark star</i>	White dwarf cools down.

Stars larger than our Sun.

Red super giant	<i>Star swells greatly</i>	Nuclear fuel begins to run out and star swells (more matter = bigger size).
Supernova	<i>Gigantic explosion due to run away fusion reactions</i>	Rapid collapse, heats to very high temperatures causing run away nuclear reactions, star explodes, flinging remnants out into space. Large gravitational forces collapse the core into a tiny space. Remains of supernova form heavier elements (Iron and above)
Neutron star	<i>Very dense star</i>	Made out of neutrons.

OR if collapse is into a really tiny space.

Black hole *No light escapes*

Gravitational forces so strong everything is pulled in.

OCR P6 Matter – models and explanations
Space – PHYSICS ONLY

Red shift

Understanding models.

Red-shift	<i>The observed increase in wavelength of light from most distant galaxies. Light moves towards the red end of the spectrum.</i>	
Hubble (1929)	<i>He studied light from distant galaxies; found as frequency decreases, wavelength increases.</i>	
	Light from star in our galaxy.	
	Light from star in nearby galaxy.	
	Light from star in distant galaxy.	
The Big Bang	<i>Universe began 13.8 billion years ago</i>	
All matter and space expanded violently from a single point.		Red—shift provides evidence for expansion.

Galaxies are moving away from us in all directions.

Light from distant galaxies is red-shifted, so galaxy is moving away from us.

Galaxies further away have bigger red-shift so are moving faster away.

Aristotle (ancient Greek)	<i>Earth at the centre, other heavenly bodies move around the Earth.</i>
Copernicus (1473 - 1543)	<i>Sun at the centre, other heavenly bodies move around the Sun.</i>
Galileo (1610)	<i>Made a telescope, looked at Jupiter, found four moons rotating around planet.</i>

Planets and moons moved at different speeds to stars = reason for different positions.